

Single-Strand Breaks Induced by Low-Energy Electrons in DNA

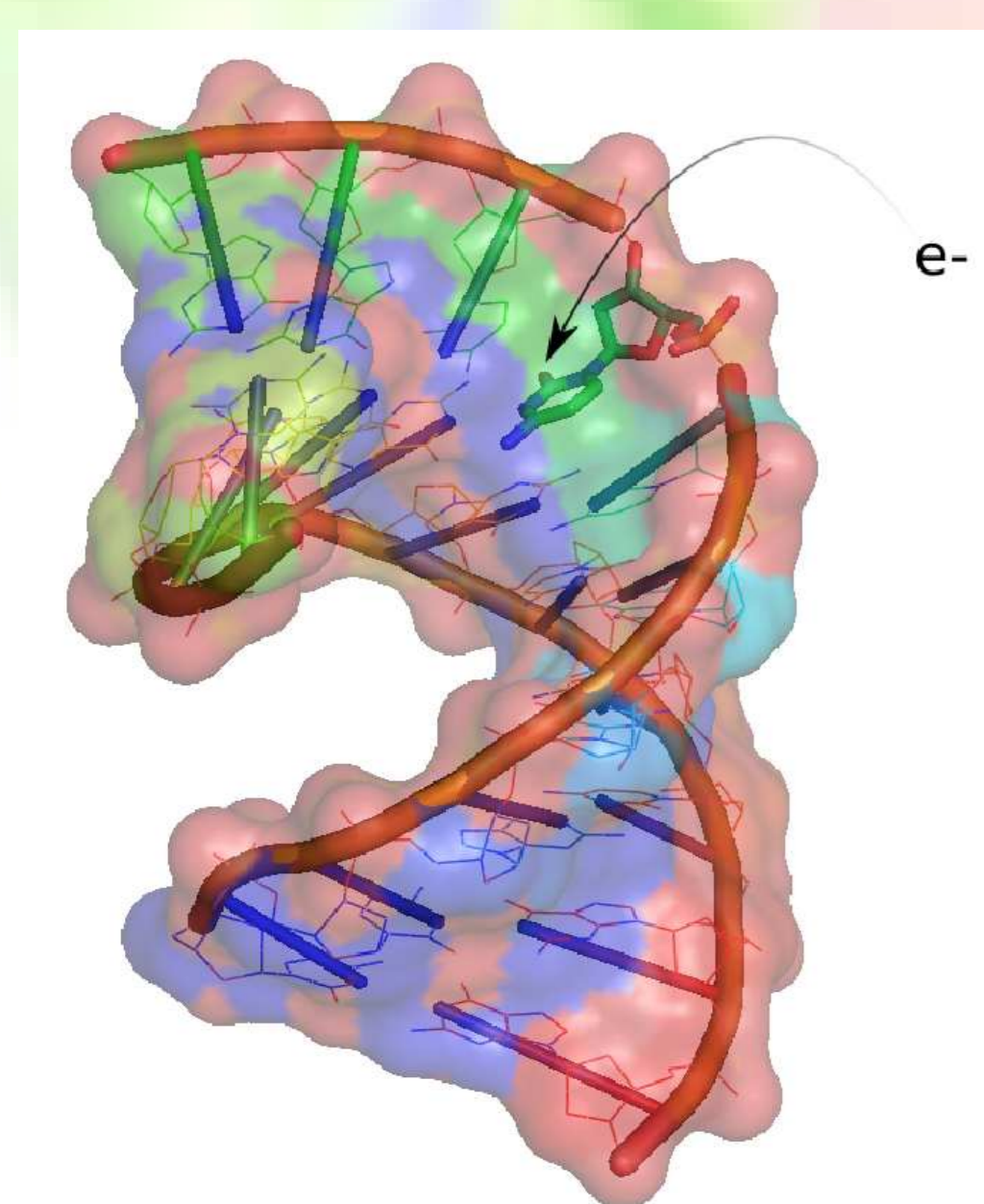
Pierre-Francois Loos and Xavier Assfeld

Equipe de Chimie et Biochimie Théoriques, UMR 7565 CNRS-UHP, Institut Jean Barriol (FR CNRS 2843),
Faculté des Sciences et Techniques, Nancy-Université, France.

Pierre-Francois.Loos@cbt.uhp-nancy.fr, Xavier.Assfeld@cbt.uhp-nancy.fr

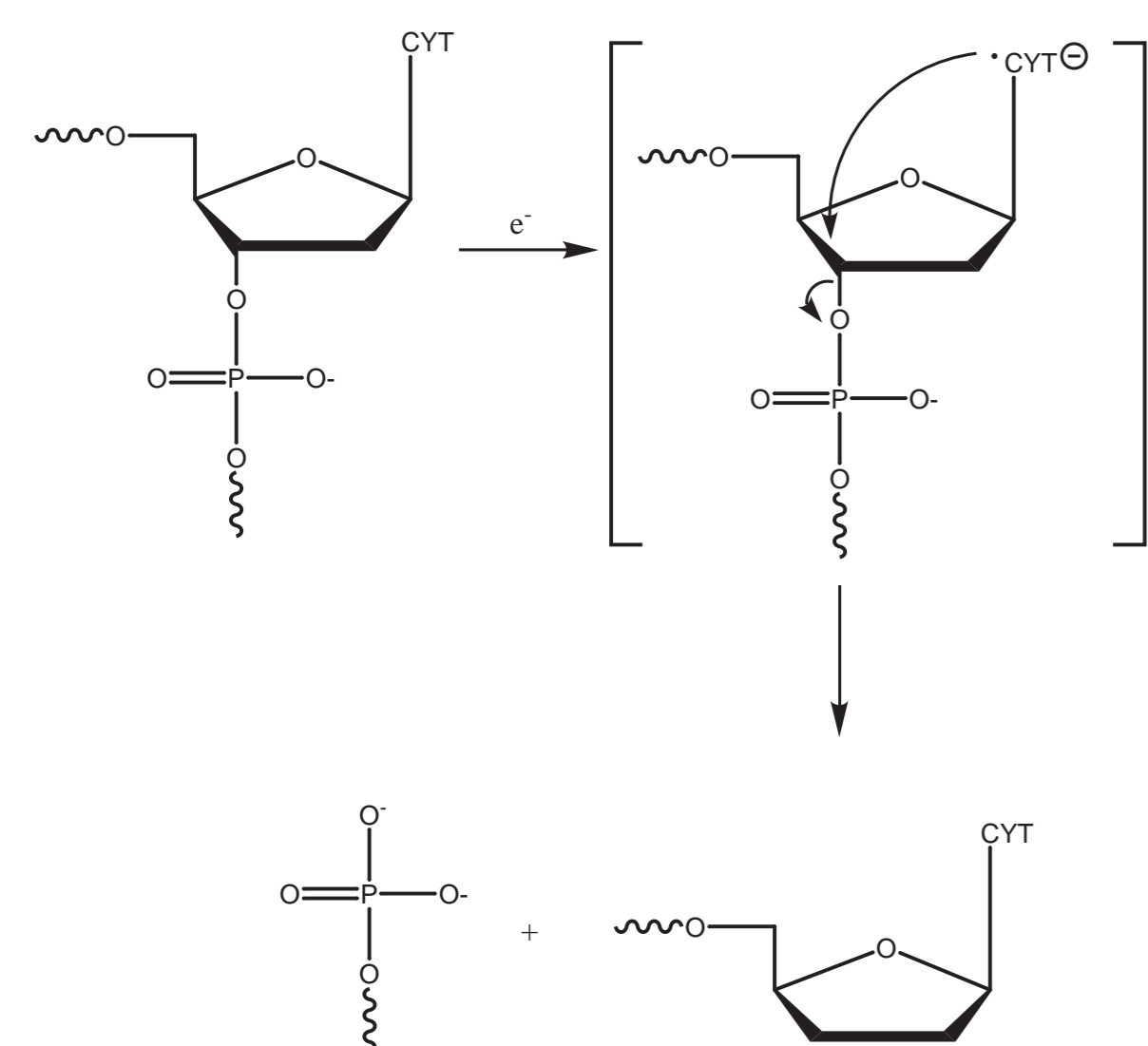
Introduction

Recent experiments have shown that low-energy electrons (0.1-2 eV) can induce single-strand breaks (SSBs) in DNA[1]. Understanding these mechanisms is of great importance for the development of medicinal sciences. Previous studies on model systems demonstrate that low-energy electrons induce the covalent-bond cleavage in DNA according to a through-bond electron transfer process between the π^* orbitals of the nucleic base and a specific bond[2, 3, 4, 5, 6, 7]. It has been shown that the C_3-O_3' σ sugar bond rupture dominates the SSBs of DNA[6, 7].



Within the LSCF/MM method[8, 9], we examine the effect of including the neighboring nucleotides at the molecular mechanic (MM) level. Because of the spatially-extended behaviour of this phenomenon, the whole nucleotide where the bond cleavage occurred must be included in the quantum mechanic (QM) part.

Proposed Mechanism of the SSB



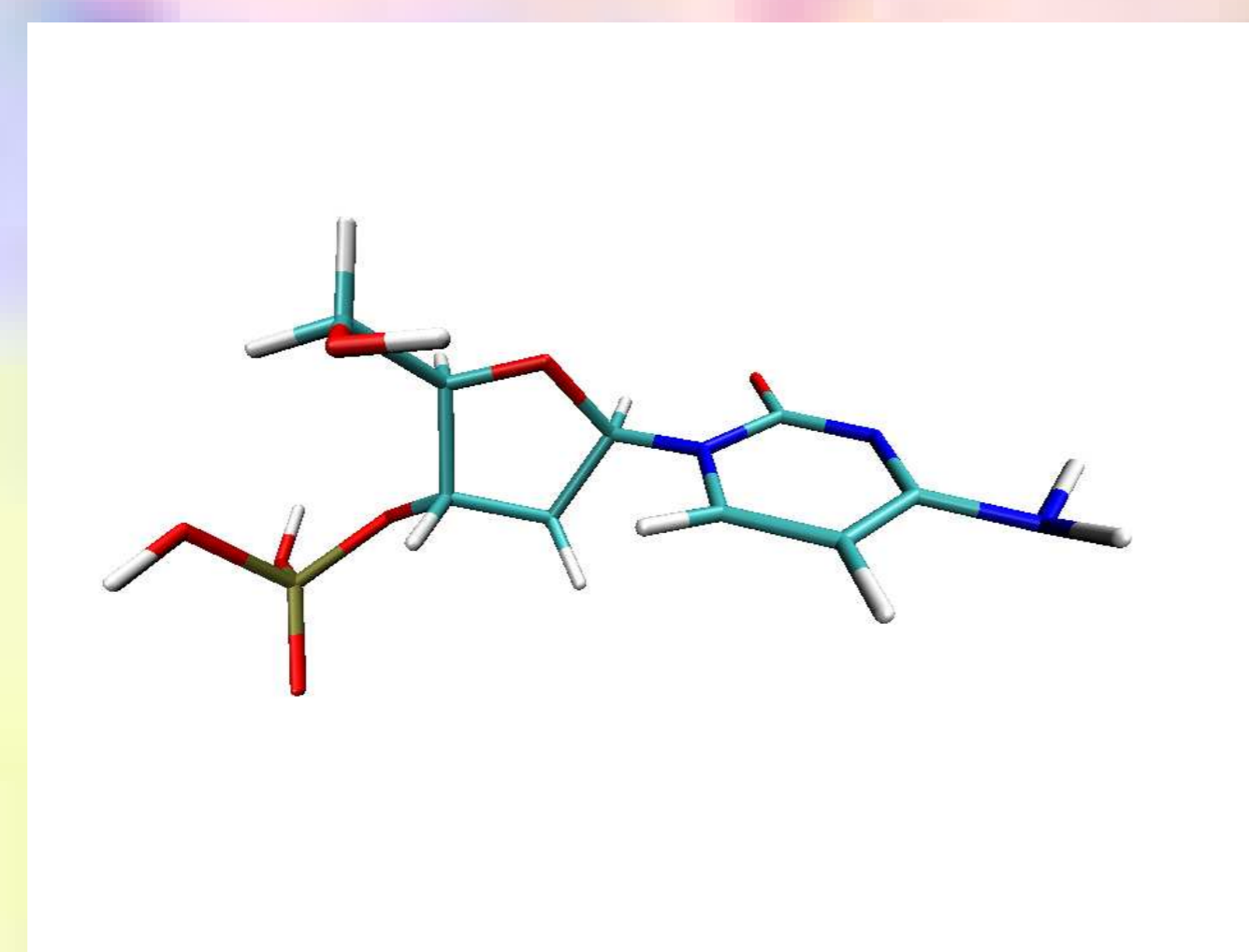
Computational details

- QM calculations : B3LYP/6-311+G(d) level of theory (Gaussian 03[10])
- MM calculations : Amber ff99 force field for DNA and RNA (Tinker4.2[11])
- SCRF calculations : PCM solvation model ($\epsilon = 4$)

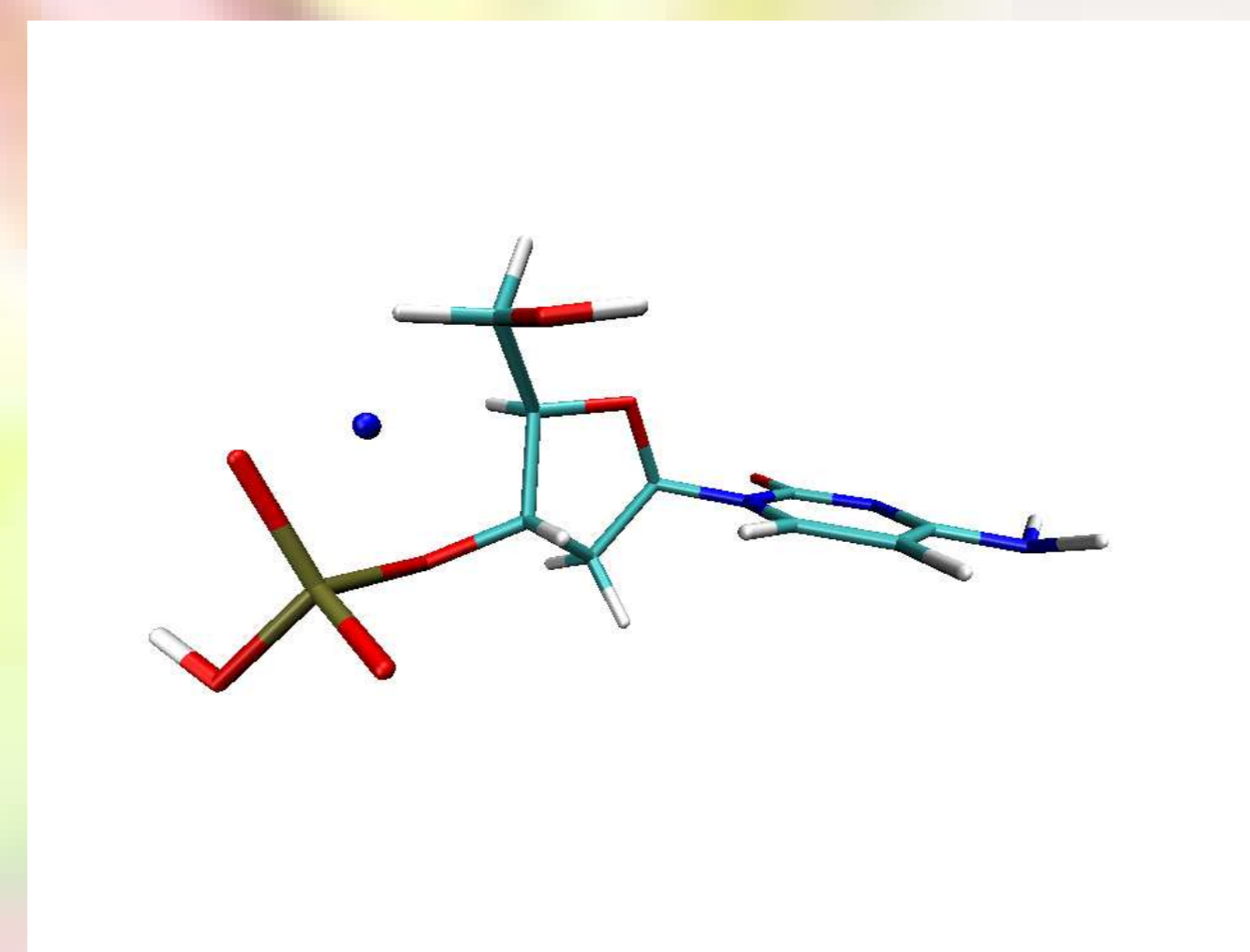
MODEL SYSTEMS

Adiabatic Electron Affinities (PCM correction)

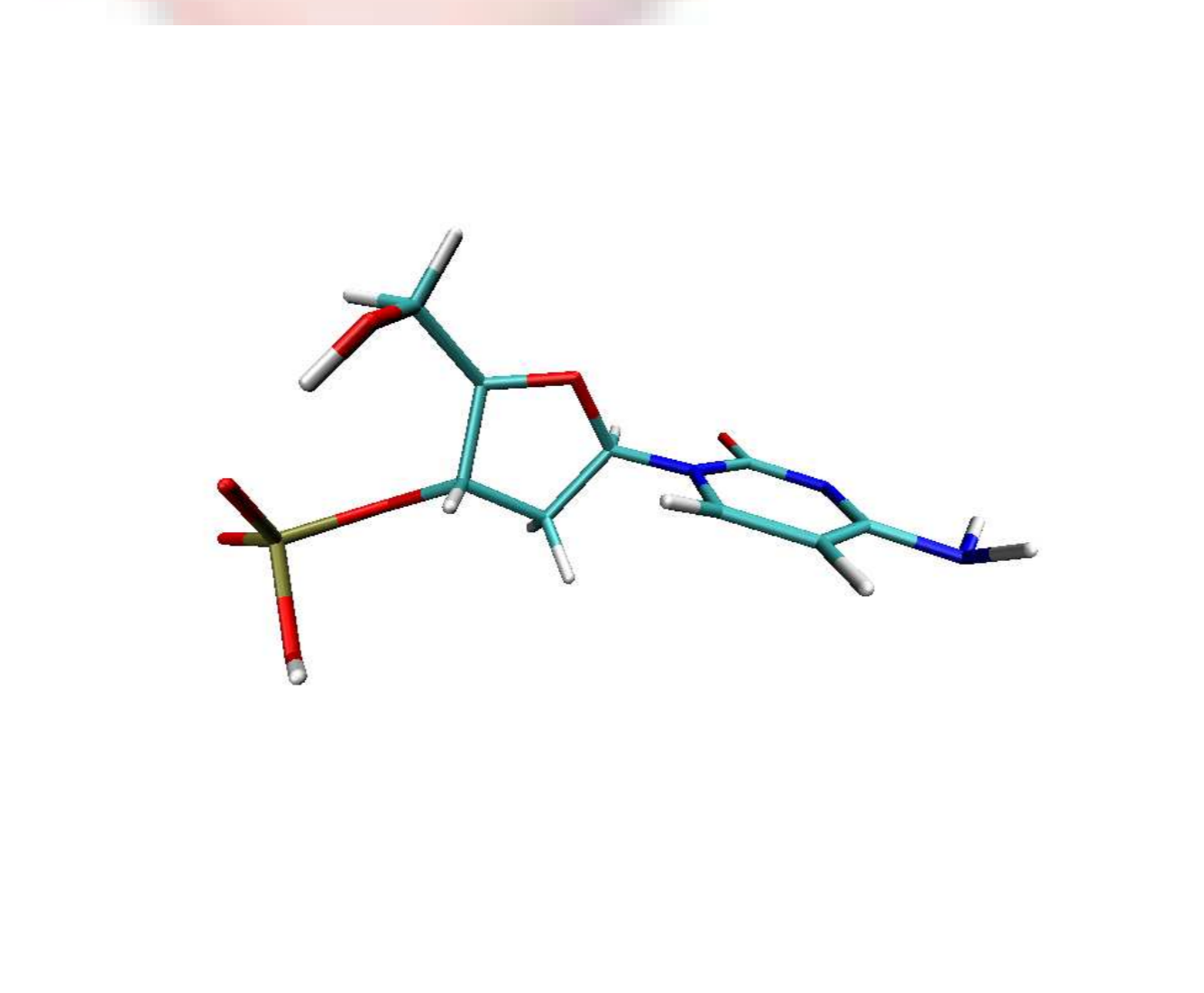
- PROTONATED FORM: 0.15 (1.51) eV



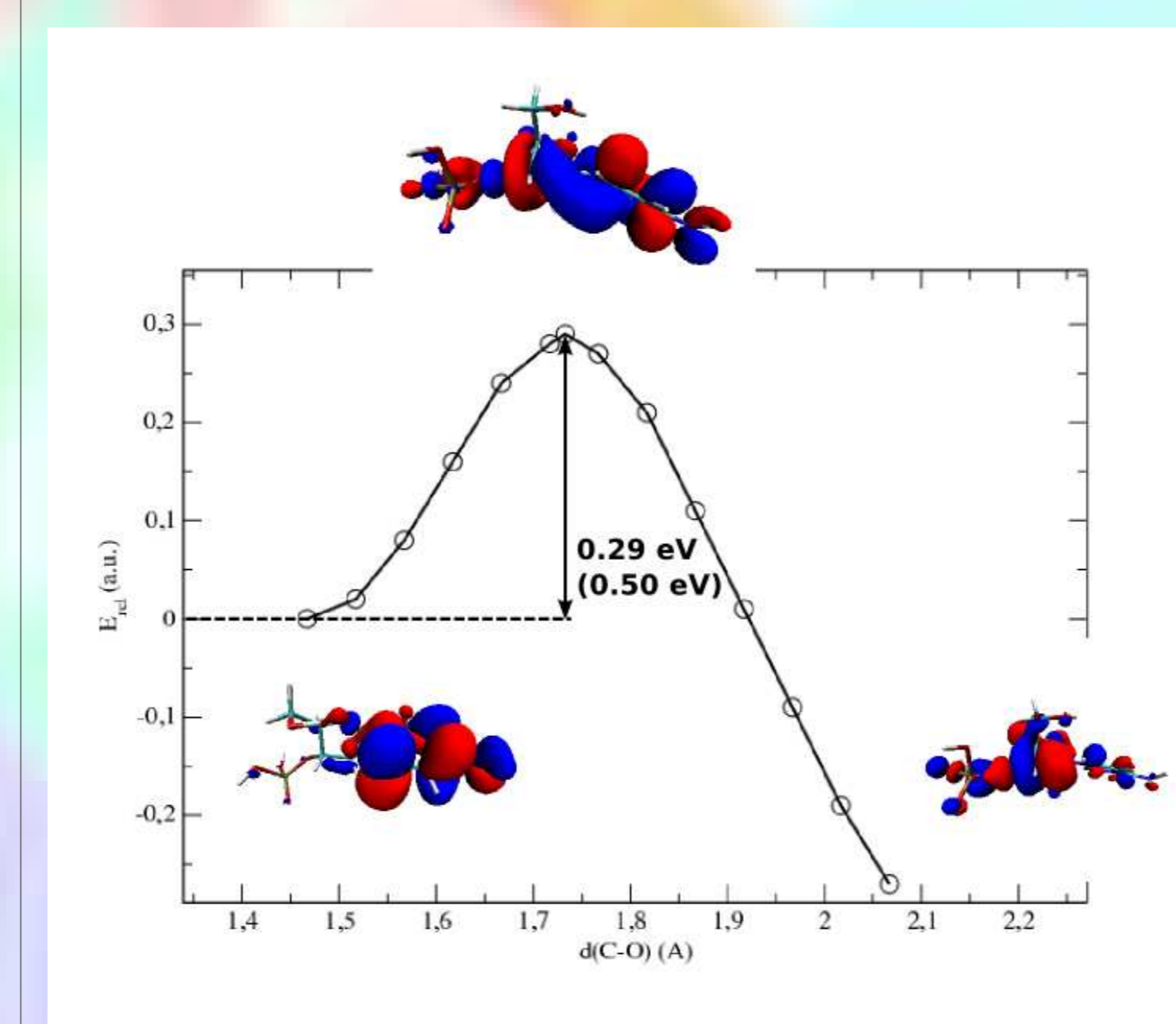
- COUNTERION (Na^+): 0.75 (1.50) eV



- DEPROTONATED FORM: -2.18 (0.94) eV



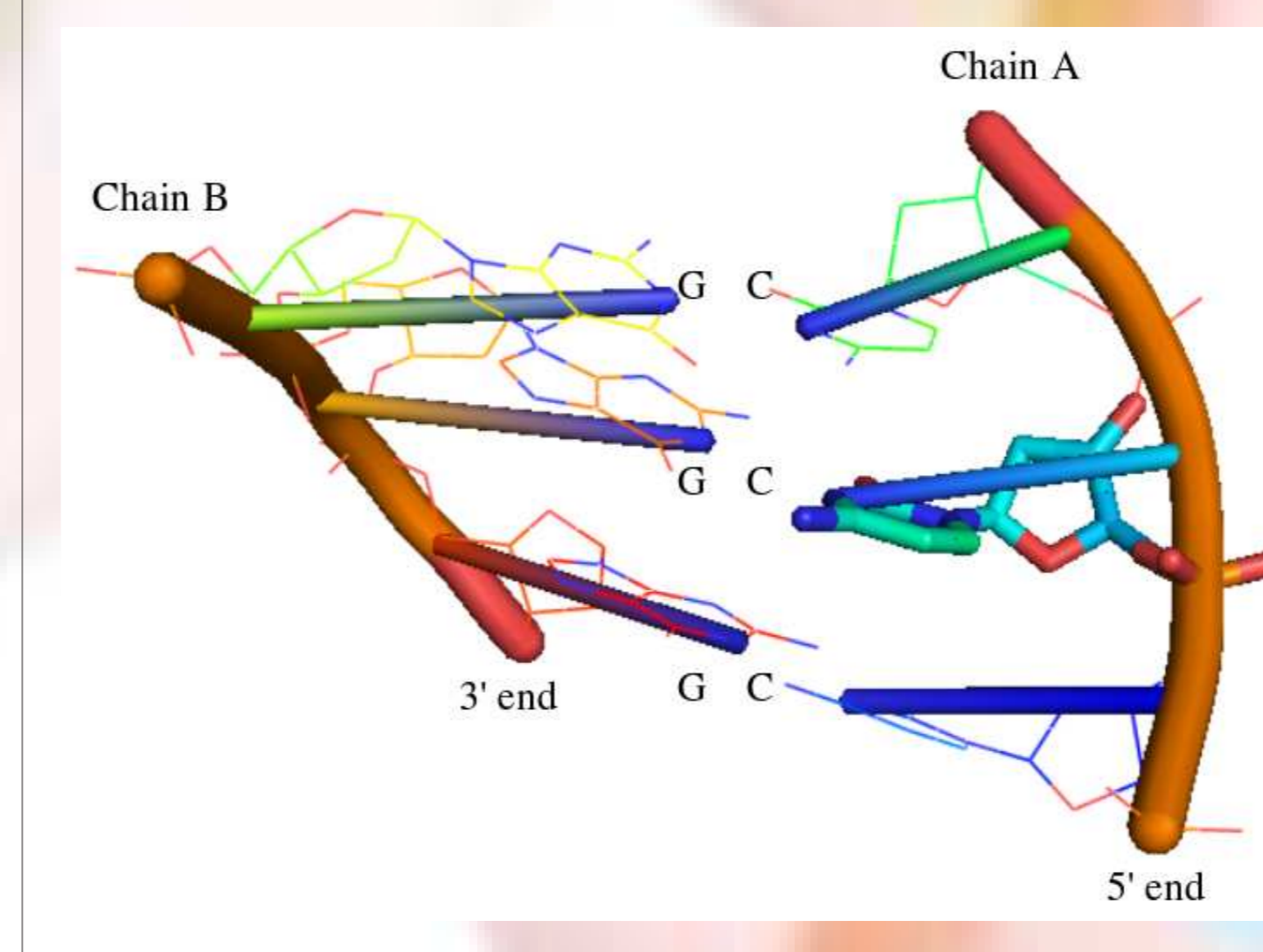
Scan of the C_3-O_3' distance



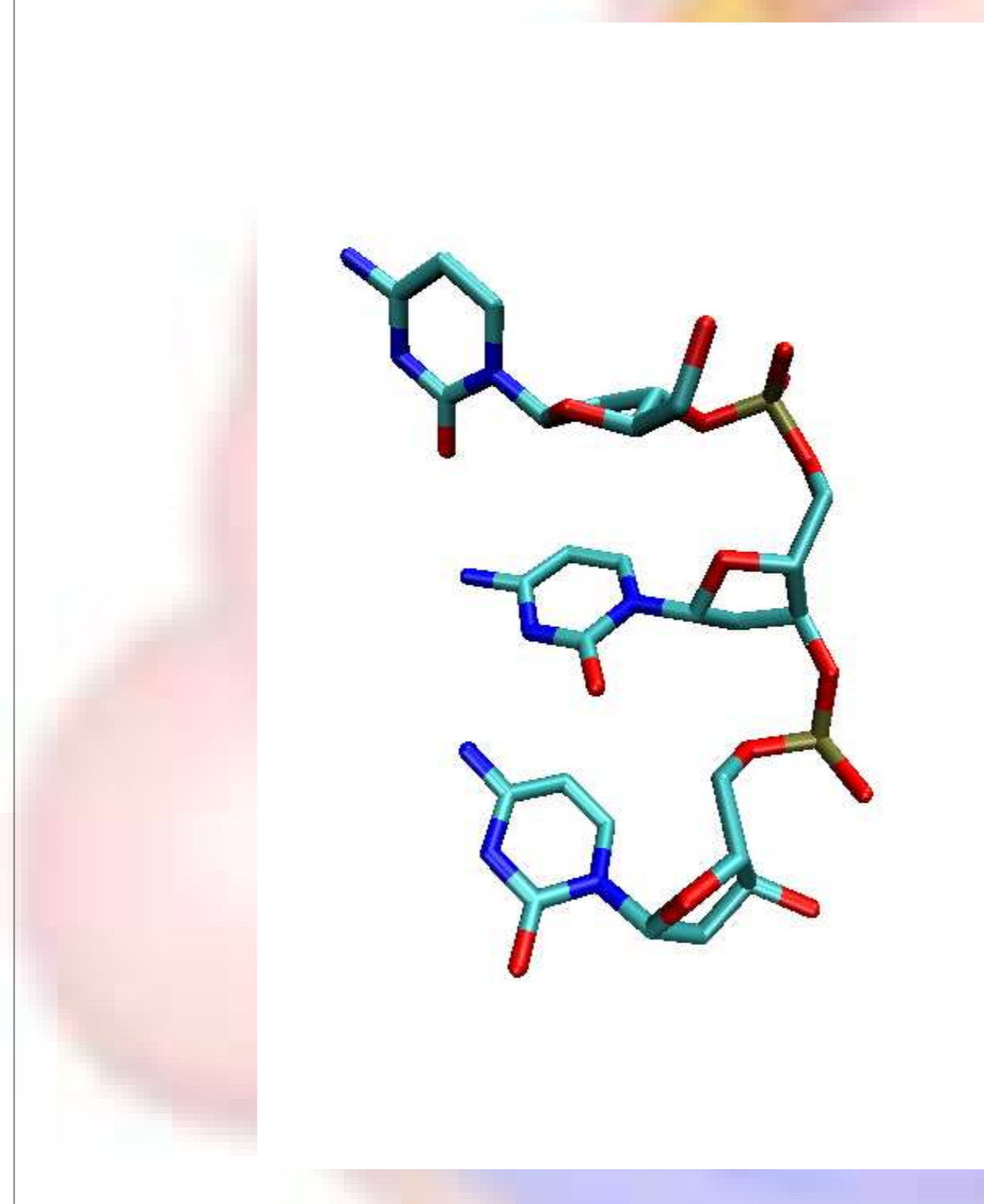
Transition state

$$d(C-O) = 1.733 \text{ \AA} (1.761 \text{ \AA})$$
$$\nu(TS) = -607 \text{ cm}^{-1} (-827 \text{ cm}^{-1})$$

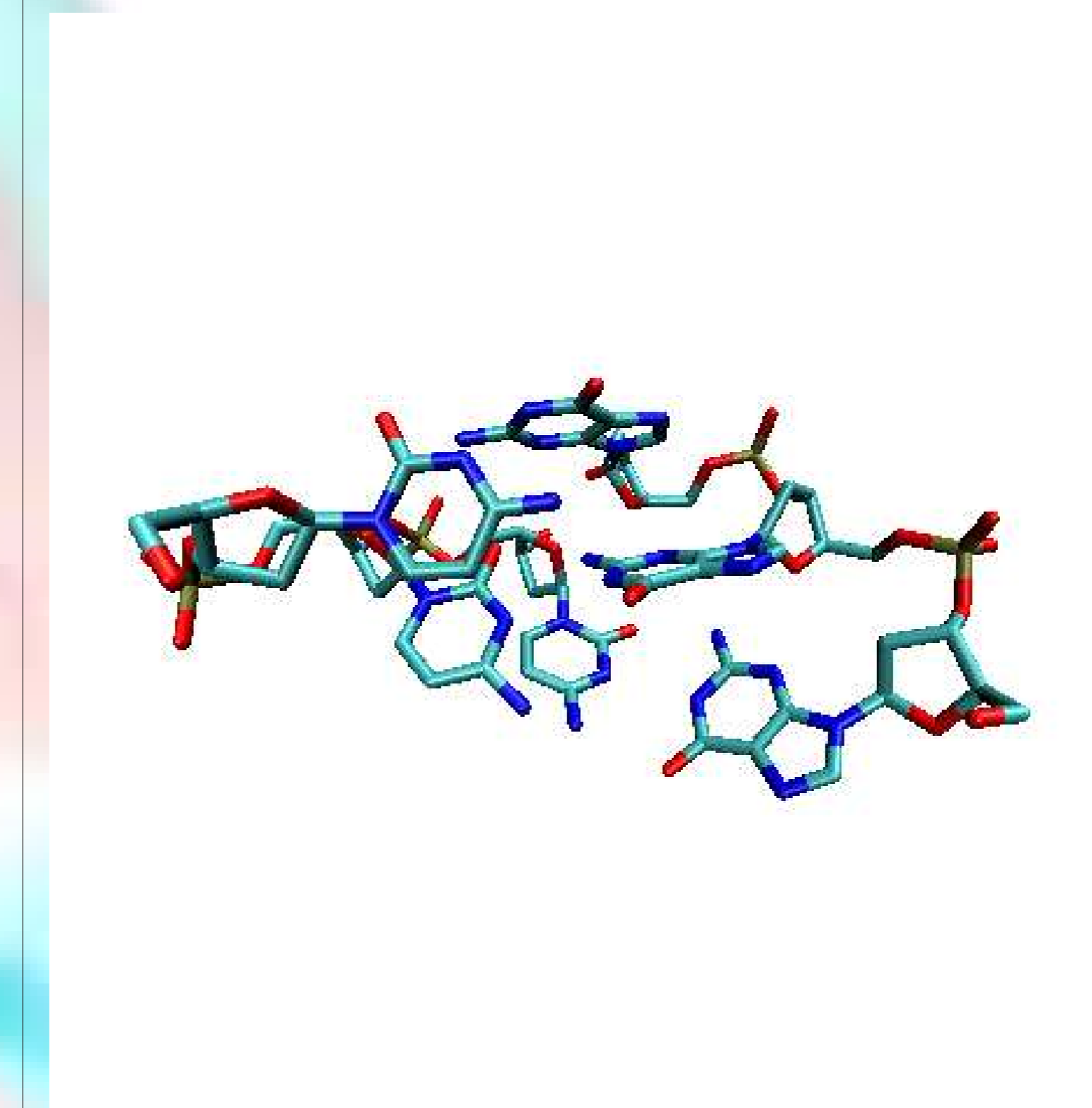
QM/MM MODELS



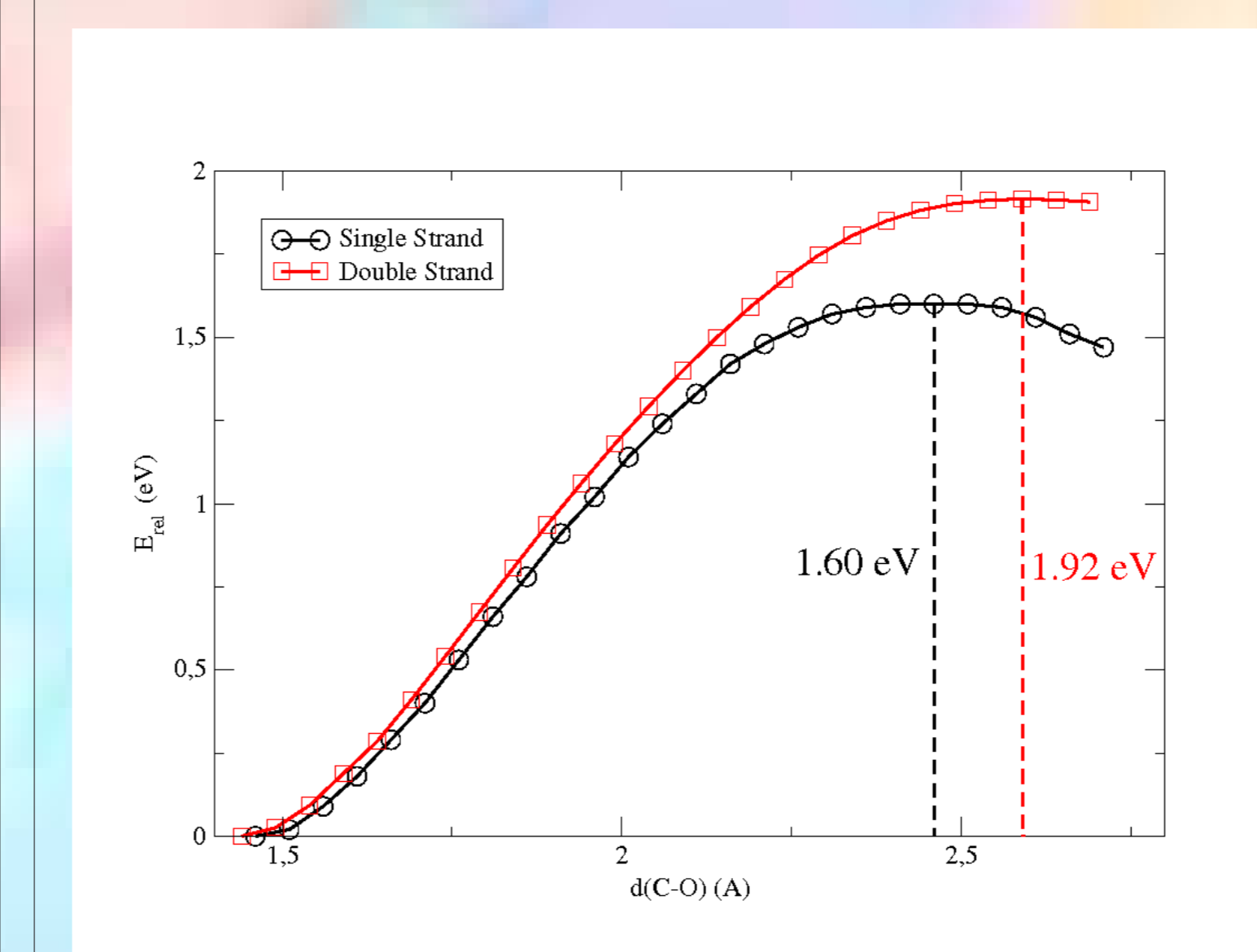
Single strand DNA



Double Strand DNA



Scan of the the C_3-O_3' bond for the Single Strand and Double Strand DNA



Stable conformations of Single Strand and Double Strand DNA

- SS: $d(C-O) = 1.46 \text{ \AA}$
- DS: $d(C-O) = 1.44 \text{ \AA}$

Transition state of Single Strand and Double Strand DNA

- SS:
 - $d(C-O) \approx 2.46 \text{ \AA}$
 - $\Delta E \approx 1.60 \text{ eV}$
- DS:
 - $d(C-O) \approx 2.59 \text{ \AA}$
 - $\Delta E \approx 1.92 \text{ eV}$

Conclusions

- Model systems:
 - SSB is independent of the phosphate protonation state
 - Surroundings effect must be taken into account
- QM/MM systems:
 - MM surroundings largely stabilize the anion (results not shown)
 - Strong effect of the complementary bases (H-bond)
 - MM surroundings increase the barrier of the C_3-O_3' bond breaking

Outlooks

- Scan of the C_3-O_3' bond for the A-DNA and B-DNA under progress
- Scan and barrier of the C_5-O_5' bond
- SSBs of GUA, ADE and THY bases

References

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