## How accurate are EOM-CC4 vertical excitation energies?

Cite as: J. Chem. Phys. 154, 221103 (2021); https://doi.org/10.1063/5.0055994
Submitted: 05 May 2021 . Accepted: 26 May 2021 . Published Online: 11 June 2021
Pierre-François Loos, Devin A. Matthews, Filippo Lipparini, and Denis Jacquemin



# How accurate are EOM-CC4 vertical excitation energies? 

Cite as: J. Chem. Phys. 154, 221103 (2021); doi: 10.1063/5.0055994
Submitted: 5 May 2021 • Accepted: 26 May 2021 •
Published Online: 11 June 2021


Export Citation
CrossMark Pierre-François Loos, ${ }^{1, a)}$ (D) Devin A. Matthews, ${ }^{2}$ (Dilippo Lipparini, ${ }^{3}$ (D) and Denis Jacquemin ${ }^{4, b)}$ (D)

AFFILIATIONS<br>${ }^{1}$ Laboratoire de Chimie et Physique Quantiques (UMR 5626), Université de Toulouse, CNRS, UPS, Toulouse, France<br>${ }^{2}$ Department of Chemistry, Southern Methodist University, Dallas, Texas 75275, USA<br>${ }^{3}$ Dipartimento di Chimica e Chimica Industriale, University of Pisa, Via Moruzzi 3, 56124 Pisa, Italy<br>${ }^{4}$ Université de Nantes, CNRS, CEISAM UMR 6230, F-44000 Nantes, France

${ }^{\text {a) }}$ Author to whom correspondence should be addressed: loos@irsamc.ups-tlse.fr
${ }^{\text {b) }}$ Electronic mail: denis.jacquemin@univ-nantes.fr


#### Abstract

We report the first investigation of the performance of EOM-CC4-an approximate equation-of-motion coupled-cluster model, which includes iterative quadruple excitations-for vertical excitation energies in molecular systems. By considering a set of 28 excited states in 10 small molecules for which we have computed CC with singles, doubles, triples, quadruples, and pentuples and full configuration interaction reference energies, we show that, in the case of excited states with a dominant contribution from the single excitations, CC4 yields excitation energies with sub-kJ mol ${ }^{-1}$ accuracy (i.e., error below 0.01 eV ), in very close agreement with its more expensive CC with singles, doubles, triples, and quadruples parent. Therefore, if one aims at high accuracy, CC4 stands as a highly competitive approximate method to model molecular excited states, with a significant improvement over both CC3 and CC with singles, doubles, and triples. Our results also evidence that, although the same qualitative conclusions hold, one cannot reach the same level of accuracy for transitions with a dominant contribution from the double excitations.


Published under an exclusive license by AIP Publishing. https://doi.org/10.1063/5.0055994

## I. INTRODUCTION

Single-reference coupled-cluster (CC) theory provides a hierarchy of size-extensive methods delivering increasingly accurate energies and properties via the systematic increase in the maximum excitation degree of the cluster operator $\hat{T}=\hat{T}_{1}+\hat{T}_{2}+\cdots+\hat{T}_{k}$ (where $k \leq n$ and $n$ is the number of electrons). ${ }^{1-5}$ Without any truncation (i.e., $k=n$ ), the so-called full CC (FCC) method is equivalent to full configuration interaction (FCI), hence providing the exact energy and wave function of the system for a given atomic basis set. However, it is not computationally viable due to its exponential scaling with system size, and one has to resort to truncated CC methods (i.e., $k<n$ ) for computational convenience. Popular choices are CC with singles and doubles (CCSD), ${ }^{1,6}$ CC with singles, doubles, and triples (CCSDT), ${ }^{7,8} \mathrm{CC}$ with singles, doubles, triples, and quadruples (CCSDTQ), ${ }^{9,10}$ and CC with singles, doubles, triples, quadruples, and pentuples (CCSDTQP) ${ }^{11,12}$ with corresponding computational scalings of $\mathcal{O}\left(N^{6}\right), \mathcal{O}\left(N^{8}\right), \mathcal{O}\left(N^{10}\right)$, and $\mathcal{O}\left(N^{12}\right)$, respectively (where $N$ denotes the number of orbitals). An alternative,
systematically improvable family of methods is defined by the $\mathrm{CC} 2,{ }^{13}$ $\mathrm{CC} 3,{ }^{14,15}$ and CC4 ${ }^{16}$ series of models, which have been introduced by the Aarhus group in the context of CC response theory. ${ }^{17}$ These iterative methods scale as $\mathcal{O}\left(N^{5}\right), \mathcal{O}\left(N^{7}\right)$, and $\mathcal{O}\left(N^{9}\right)$, respectively, and can be seen as cheaper approximations of CCSD, CCSDT, and CCSDTQ, by skipping the most expensive terms and avoiding the storage of the higher-excitation amplitudes. A somewhat similar strategy has been applied to define the CCSDT-3 ${ }^{18,19}$ and CCSDTQ-3 ${ }^{16}$ models based on arguments stemming from perturbation theory. Of course, a large number of other approximate CC models have been developed over the years and we refer the interested reader to specialized reviews for more details. ${ }^{3-5,20}$

Coupled-cluster methods have been particularly successful for small- and medium-sized molecules in the field of thermodynamics, kinetics, and spectroscopy, thanks to the computations of accurate equilibrium geometries, ${ }^{21}$ potential energy surfaces, vibrational frequencies, ${ }^{22}$ Born-Oppenheimer corrections, ${ }^{23}$ and a vast panel of properties, such as dipoles (and higher moments), ${ }^{21}$ nuclear magnetic resonance (NMR) chemical shifts, ${ }^{22}$ magnetizabilities, ${ }^{24}$
and polarizabilities. ${ }^{25}$ Although originally developed for groundstate energies and properties, CC has been successfully extended to excited states ${ }^{26}$ thanks to the equation-of-motion (EOM) ${ }^{27-30}$ and linear-response (LR) ${ }^{31-35}$ formalisms, which are known to produce identical excitation energies but different properties. In EOM-CC, one determines vertical excitation energies via the diagonalization of the similarity-transformed Hamiltonian $\bar{H}=e^{-\hat{T}} \hat{H} e^{\hat{T}}$. A more general procedure to compute excitation energies that can be applied to any approximate CC model (as the ones mentioned above) consists in diagonalizing the so-called CC Jacobian obtained via the differentiation of the CC amplitude equations. Again, by increasing the maximum excitation degree of $\hat{T}$, one can systematically produce increasingly accurate EOM/LR excitation energies with the "complete" series CCSD, ${ }^{29}$ CCSDT, ${ }^{36,37}$ CCSDTQ, ${ }^{38}$ CCSDTQP, ${ }^{39}$ etc. or the "approximate" series CC2, ${ }^{13} \mathrm{CC} 3,{ }^{14} \mathrm{CC} 4,{ }^{39}$ etc. while retaining the formal scaling of their ground-state analog. From here on, we drop the EOM prefix as all our calculations are performed within this approach.

Recently, a large collection of deterministic, stochastic, or hybrid selected CI (SCI) methods ${ }^{40-42}$ has (re)appeared ${ }^{43-57}$ in the electronic structure landscape providing an alternative route to highly accurate ground- and excited-state energies ${ }^{5-70}$ (see Refs. 71 and 72 for recent reviews). The general idea behind SCI methods is simple: rather than exploring the entire FCI space by systematically increasing the maximum excitation degree of the determinants taken into account (leading to the slowly convergent and sizeinconsistent series of methods CISD, CISDT, CISDTQ, etc.), one performs a sparse exploration of the FCI space by selecting only the most energetically important determinants thanks to a suitable criterion usually based on perturbation theory. ${ }^{42,67,73,74}$ By iteratively increasing the number of determinants of the variational space and supplementing it with a second-order perturbative correction (PT2), the SCI + PT2 family of methods has been recently shown to produce near-FCI correlation and excitation energies for small- and medium-size molecules in compact basis sets. ${ }^{58,61-65,67,70}$ Although the formal scaling of such algorithms remains exponential, the prefactor is greatly reduced, which explains their current attractiveness in the electronic structure community and much wider applicability than their standard FCI parent.

Taking advantage of the high accuracy of CC and SCI+PT2 methods, we have very recently created a large dataset gathering more than 500 highly accurate vertical excitation energies for electronic transitions of various natures (valence, Rydberg, $n \rightarrow \pi^{*}$, $\pi \rightarrow \pi^{*}$, singlet, doublet, triplet, charge-transfer, and double excitations) in small- and medium-sized molecules ranging from diatomics to molecules as large as naphthalene. ${ }^{61,63-65,70,71}$ The main purpose of this so-called QUEST database is to provide reference excitation energies in order to perform fair and reliable benchmarks between electronic structure methods and assess their strengths and weaknesses for a large panel of chemical scenarios. Most of these reference transition energies, which rely exclusively on high-level $a b$ initio calculations, can be reasonably considered as chemically accurate, i.e., within $1 \mathrm{kcal} \mathrm{mol}^{-1}$ or 0.043 eV of the FCI limit. However, their accuracy may rapidly deteriorate, in particular, as the system size grows. Indeed, it is usually challenging to compute reliable SCI+PT2 estimates or CCSDTQ excitation energies for molecules with more than four non-hydrogen atoms. Therefore, for the larger
molecules of the QUEST database, we mostly relied on CCSDT to define reference excitation energies. Of course, it would be highly valuable to have access to more accurate methods (including, at least, quadruple excitations) in order to refine these theoretical best estimates.

In this context, the main purpose of the present study is to assess the relative accuracy of the approximate iterative CC4 model against the more expensive CCSDTQ and CCSDTQP methods in the case of vertical excitation energies, as well as their absolute accuracy with respect to FCI. To do so, we consider a set of ten small molecules $\left(\mathrm{NH}_{3}, \mathrm{C}_{2}, \mathrm{BH}, \mathrm{BF}, \mathrm{CO}, \mathrm{N}_{2}, \mathrm{HCl}, \mathrm{H}_{2} \mathrm{~S}, \mathrm{HNO}\right.$, and $\mathrm{H}_{2} \mathrm{O}$ ) and we compare the excitation energies associated with 28 singlet excited states of various natures ( $n \rightarrow \pi^{*}, \pi \rightarrow \pi^{*}$, Rydberg, valence, charge-transfer, and double excitations) and spatial symmetries obtained with various high-level CC methods. Although a small number of studies have been published on the performance of CC4 for ground-state energies and properties, ${ }^{16,75}$ this work stands, to the best of our knowledge, as the first to consider CC4 for the computation of excited-state energies. As we shall see below, CC4 is an excellent approximation to its CCSDTQ parent and produces, in the case of excited states with a dominant contribution from the single excitations, excitation energies with sub- $\mathrm{kJ} \mathrm{mol}^{-1}$ accuracy (i.e., error below 0.01 eV ) for this set of small molecules, well below the chemical accuracy threshold.

## II. COMPUTATIONAL DETAILS

All the methods considered in the present study are listed in Table I alongside their formal computational scaling and the electronic structure software employed to compute the excitation energies. In a nutshell, we have used by default CFOUR ${ }^{76}$ to compute the CC energies at the notable exception of CCSDTQP for which we have employed MRCC and its automated implementation of highorder CC methods. ${ }^{77}$ The present CFOUR calculations have been performed with the new and fast CC module (xncc) written by one of the authors (DAM), which couples a general algebraic and graphical interpretation of the non-orthogonal spin-adaptation approach with highly efficient storage format and implementation techniques designed to minimize data movement and to avoid costly tensor transposes. ${ }^{78}$ The FCI estimates were obtained with the SCI algorithm known as "configuration interaction using a perturbative selection made iteratively" (CIPSI) implemented in QUANTUM PACKAGE. ${ }^{67}$ The error bars associated with the extrapolation step of the

TABLE I. Methods considered in the present study, their formal computational scaling, and the electronic structure software employed to compute excitation energies. Here, $N$ is the number of orbitals.

| Method | Scaling | Code | References |
| :--- | :---: | :---: | :---: |
| CC2 | $\mathcal{O}\left(N^{5}\right)$ | CFOUR | 76 |
| CCSD | $\mathcal{O}\left(N^{6}\right)$ | CFOUR | 76 |
| CC3 | $\mathcal{O}\left(N^{7}\right)$ | CFOUR | 76 |
| CCSDT | $\mathcal{O}\left(N^{8}\right)$ | CFOUR | 76 |
| CC4 | $\mathcal{O}\left(N^{9}\right)$ | CFOUR | 76 |
| CCSDTQ | $\mathcal{O}\left(N^{10}\right)$ | CFOUR | 76 |
| CCSDTQP | $\mathcal{O}\left(N^{12}\right)$ | MRCC | 77 |
| CIPSI | $\mathcal{O}\left(e^{N}\right)$ | QUANTUM PACKAGE | 67 |

CIPSI calculations (see Ref. 67) have been computed using our recently developed protocol presented in Ref. 70.

Because this is, to our knowledge, the first implementation of EOM-CC4, we have verified its accuracy by comparing the excitation energies obtained from solving for the right- and left-hand wave functions. Coupled cluster is a non-Hermitian theory, and thus, the right- and left-hand eigenfunctions of the Jacobian are distinct, albeit with the same eigenvalue. For the left-hand solution, we have reused the already-verified code for the ground-state $\Lambda$ equations, which describe the amplitude relaxation contribution in the analytic gradient theory. ${ }^{75}$ The structures of the left-hand EOM-CC and $\Lambda$ equations are identical, and so simply interfacing this existing code with a Davidson solver ${ }^{79}$ provides left-hand EOM-CC solutions; this procedure has been checked for other known-good methods, such as CCSDT and CC3.

All calculations have been performed in the frozen-core approximation, and the CC3/aug-cc-pVTZ geometries of the systems considered here have been extracted from previous studies. ${ }^{61,80}$ In the following, we consider diffuse-containing Dunning's doubleand triple- $\zeta$ basis sets (aug-cc-pVDZ and aug-cc-pVTZ). Note that CCSDTQP energies could only be computed for the smaller basis (aug-cc-pVDZ).

## III. RESULTS AND DISCUSSION

Table II gathers, for the two considered basis sets, 25 vertical excitation energies with a (strongly) dominant contribution from the single excitations computed for a set of eight molecules with various CC models as well as the FCI estimates computed with CIPSI. First, we underline that the FCI estimates show how accurate the CCSDTQP reference data are, with a maximum deviation of 0.002 eV when one considers the aug-cc-pVDZ basis. For the larger
aug-cc-pVTZ basis, the CCSDTQ and FCI remain in excellent agreement although the error bars associated with the extrapolated FCI values prevent us from any quantitative comparisons.

The mean absolute errors (MAEs) and mean signed errors (MSEs) with respect to CCSDTQP computed in the aug-cc-pVDZ basis are reported in the bottom of Table II for the CC2-to-CCSDTQ models. The distribution of the errors is reported in Fig. 1 for each level of theory. These statistical quantities nicely illustrate the systematic improvement of the transition energies when one ramps up the computational effort following the series CC2, CCSD, CC3, CCSDT, CC4, and CCSDTQ. We note that the errors decrease by roughly one order of magnitude when switching from CCSD to CC3 and from CCSDT to CC4, while improvements of $\sim 50 \%$ "only" are noted when going from the "approximate" model to the "complete" method (i.e., from CC2 to CCSD, from CC3 to CCSDT, and from CC4 to CCSDTQ). In other words, CC4 brings significant improvements in terms of MAE and MSE as compared to the third-order methods, CC3 and CCSDT, which demonstrates the importance of quadruple excitations when one aims at very high accuracy. Besides, for the two basis sets, there is an outstanding similarity between the CC4 and CCSDTQ excitation energies with mean absolute and signed deviations below (equal to) 0.001 eV and a maximum deviation of $0.011 \mathrm{eV}(0.007 \mathrm{eV})$ between the two sets of data obtained with the aug-cc-pVDZ (aug-cc-pVTZ) basis set. Therefore, including quadruples allows us to reach sub- $\mathrm{kJ} \mathrm{mol}^{-1}$ accuracy (i.e., average error below 0.01 eV ) for transitions dominated by single excitations with only a rather minor improvement in going from CC 4 to CCSDTQ.

A closer inspection at Table II shows that the largest deviations appear for the transitions with the smallest $\% T_{1}$ values (where $\% T_{1}$ is the percentage of single excitations involved in the transition, which is computed at the CC3/aug-cc-pVTZ level in the present case). This is particularly noticeable for the two ${ }^{1} \Pi_{u}$ transitions of $\mathrm{N}_{2}$. This


FIG. 1. Distribution of the error (in eV) in excitation energies (with respect to CCSDTQP) for CC2, CC3, and CC4 (top) and CCSD, CCSDT, and CCSDTQ (bottom) obtained with the aug-cc-pVDZ basis. Note the various error ranges ( $x$ axis) for the various methods. See Table II for the raw data.

TABLE III. Vertical excitation energies (in eV ) for a selection of transitions with a dominant contribution from the double excitations obtained at various levels of theory with the aug-cc-pVDZ basis set. $\% T_{1}$ is the percentage of single excitations involved in the transition (computed at the CC3/aug-cc-pVTZ level).

| Mol. | State | $\% T_{1}$ | CC3 | CCSDT | CC4 | CCSDTQ | CCSDTQP | FCI |
| :--- | :--- | :---: | :---: | :---: | :---: | :---: | :---: | ---: |
| $\mathrm{C}_{2}$ | ${ }^{1} \Delta_{g}$ | 1 | 3.107 | 2.632 | 2.341 | 2.241 | 2.214 | $2.213(0)$ |
|  | ${ }^{1} \Sigma_{g}^{+}$ | 1 | 3.283 | 2.874 | 2.602 | 2.521 | 2.505 | $2.503(1)$ |
| HNO | ${ }^{1} A^{\prime}$ | 0 | 5.247 | 4.756 | 4.454 | 4.424 |  | $4.397(1)$ |



FIG. 2. Error in excitation energies (with respect to FCl ) computed with CC 3 , CCSDT, CC4, and CCSDTQ in the aug-cc-pVDZ basis for transitions with a dominant contribution from the double excitations gathered in Table III.
suggests that the performance of the various CC models discussed above might be highly dependent on the nature of the transitions, as discussed in Ref. 63. To investigate further this point, we report in Table III vertical excitation energies for transitions with a dominant contribution from the doubly excited determinants in the carbon dimer, $\mathrm{C}_{2}$, and nitroxyl, HNO, computed with the aug-cc-pVDZ basis for the methods including at least triple excitations; the secondorder methods, CC2 and CCSD, are unable to faithfully locate these states with a large contribution of double excitations. These transitions can be labeled as "pure" double excitations as they involve an insignificant amount of single excitations ( $\% T_{1} \approx 0$ ), hence providing a very stringent test for the EOM-CC formalism. In this case, as shown in Fig. 2, the differences between methods are magnified, but the conclusions drawn in the previous paragraph hold: CC4 is an excellent approximation to CCSDTQ (with a maximum deviation of 0.1 eV for the ${ }^{1} \Delta_{g}$ transition of $\mathrm{C}_{2}$ ) and a massive improvement over CC3 (where the error can be as large as 0.9 eV ) and, to a lesser extent, over CCSDT. However, for these transitions with a dominant double excitation character, CC4 does not permit us to reach chemical accuracy with errors of the order of 0.1 eV compared to CCSDTQP and FCI. The outcome might differ for transitions of mixed characters ( $\% T_{1} \approx 70$ ) such as the well-known ${ }^{1} A_{g}$ excited state of butadiene. ${ }^{63,81-85}$

## IV. CONCLUSION

Thanks to the results gathered in the present study, we can conclude, for this set of small molecules at least, that CC4 is a rather competitive approximation to its more expensive CCSDTQ parent as well as a very significant improvement over both its third-order version, CC3, and the "complete" CCSDT method. This is particularly true in the case of transitions with a dominant contribution from the single excitations (Table II) when one reaches sub-kJ mol ${ }^{-1}$
accuracy. For states with a dominant contribution from the double excitations, we have seen (Table III) that the same qualitative conclusions hold but one cannot reach chemical accuracy for the set of "pure" double excitations that we have considered.

These findings are promising, though we are well aware that the conclusions obtained for small and larger molecules might differ significantly. For example, CCSD outperforms CC2 for compact molecules (as here), but the opposite trend is often found for larger compounds. ${ }^{61,70,71}$ Therefore, although further investigations on larger compounds are definitely required, the present results are very encouraging as CC4, with its $\mathcal{O}\left(N^{9}\right)$ scaling, can be applied to significantly larger molecules than CCSDTQ [which scales as $\mathcal{O}\left(N^{10}\right)$ ]. This will likely allow us to revisit, in the future, some of the theoretical best estimates defined in the QUEST database. ${ }^{70,71}$

## ACKNOWLEDGMENTS

P.-F.L. has received funding from the European Research Council (ERC) under the European Union's Horizon 2020 research and innovation program (Grant Agreement No. 863481). D.A.M. has received funding from the National Science Foundation (Grant No. OAC-2003931). This work was performed using HPC resources from CALMIP (Toulouse) under Allocation No. 2021-18005 and of the CCIPL computational center installed in Nantes.

## DATA AVAILABILITY

The data that support the findings of this study are openly available in Zenodo at http://doi.org/10.5281/zenodo.4739288, Ref. 86.

## REFERENCES

${ }^{1}$ J. Čížek, J. Chem. Phys. 45, 4256 (1966).
${ }^{2}$ J. Paldus, J. Čížek, and I. Shavitt, Phys. Rev. A 5, 50 (1972).
${ }^{3}$ T. D. Crawford and H. F. Schaefer, Reviews in Computational Chemistry (John Wiley \& Sons, Ltd., 2000), pp. 33-136.
${ }^{4}$ R. J. Bartlett and M. Musiał, Rev. Mod. Phys. 79, 291 (2007).
${ }^{5}$ I. Shavitt and R. J. Bartlett, Many-Body Methods in Chemistry and Physics: MBPT and Coupled-Cluster Theory, Cambridge Molecular Science (Cambridge University Press, Cambridge, 2009).
${ }^{6}$ G. D. Purvis and R. J. Bartlett, J. Chem. Phys. 76, 1910 (1982).
${ }^{7}$ J. Noga and R. J. Bartlett, J. Chem. Phys. 86, 7041 (1987).
${ }^{8}$ G. E. Scuseria and H. F. Schaefer, Chem. Phys. Lett. 152, 382 (1988).
${ }^{9}$ N. Oliphant and L. Adamowicz, J. Chem. Phys. 95, 6645 (1991).
${ }^{10}$ S. A. Kucharski and R. J. Bartlett, J. Chem. Phys. 97, 4282 (1992).
${ }^{11}$ S. Hirata and R. J. Bartlett, Chem. Phys. Lett. 321, 216 (2000).
${ }^{12}$ M. Kállay and P. R. Surján, J. Chem. Phys. 115, 2945 (2001).
${ }^{13}$ O. Christiansen, H. Koch, and P. Jørgensen, Chem. Phys. Lett. 243, 409 (1995).
${ }^{14}$ O. Christiansen, H. Koch, and P. Jørgensen, J. Chem. Phys. 103, 7429 (1995).
${ }^{15}$ H. Koch, O. Christiansen, P. Jørgensen, A. M. Sanchez de Merás, and T. Helgaker, J. Chem. Phys. 106, 1808 (1997).
${ }^{16}$ M. Kállay and J. Gauss, J. Chem. Phys. 123, 214105 (2005).
${ }^{17}$ O. Christiansen, P. Jørgensen, and C. Hättig, Int. J. Quantum Chem. 68, 1 (1998).
${ }^{18}$ M. Urban, J. Noga, S. J. Cole, and R. J. Bartlett, J. Chem. Phys. 83, 4041 (1985).
${ }^{19}$ J. Noga, R. J. Bartlett, and M. Urban, Chem. Phys. Lett. 134, 126 (1987).
${ }^{20}$ P. Piecuch, K. Kowalski, I. S. O. Pimienta, and M. J. Mcguire, Int. Rev. Phys. Chem. 21, 527 (2002).
${ }^{21}$ M. Kállay, J. Gauss, and P. G. Szalay, J. Chem. Phys. 119, 2991 (2003).
${ }^{22}$ M. Kállay and J. Gauss, J. Chem. Phys. 120, 6841 (2004).
${ }^{23}$ J. Gauss, A. Tajti, M. Kállay, J. F. Stanton, and P. G. Szalay, J. Chem. Phys. 125, 144111 (2006).
${ }^{24}$ J. Gauss, M. Kállay, and F. Neese, J. Phys. Chem. A 113, 11541 (2009).
${ }^{25}$ M. Kállay and J. Gauss, J. Mol. Struct.: THEOCHEM 768, 71 (2006).
${ }^{26}$ H. J. Monkhorst, Int. J. Quantum Chem. 12, 421 (1977).
${ }^{27}$ K. Emrich, Nucl. Phys. A 351, 379 (1981).
${ }^{28}$ D. C. Comeau and R. J. Bartlett, Chem. Phys. Lett. 207, 414 (1993).
${ }^{29}$ J. F. Stanton and R. J. Bartlett, J. Chem. Phys. 98, 7029 (1993).
${ }^{30}$ A. I. Krylov, Annu. Rev. Phys. Chem. 59, 433 (2008).
${ }^{31}$ S. Ghosh, D. Mukherjee, and S. Bhattacharyya, Mol. Phys. 43, 173 (1981).
${ }^{32}$ H. Sekino and R. J. Bartlett, Int. J. Quantum Chem. 26, 255 (1984).
${ }^{33}$ H. Koch and P. Jørgensen, J. Chem. Phys. 93, 3333 (1990).
${ }^{34}$ H. Koch, H. J. A. Jensen, P. Jørgensen, and T. Helgaker, J. Chem. Phys. 93, 3345 (1990).
${ }^{35}$ R. J. Rico and M. Head-Gordon, Chem. Phys. Lett. 213, 224 (1993).
${ }^{36}$ K. Kowalski and P. Piecuch, J. Chem. Phys. 115, 643 (2001).
${ }^{37}$ S. A. Kucharski, M. Włoch, M. Musiał, and R. J. Bartlett, J. Chem. Phys. 115, 8263 (2001).
${ }^{38}$ S. Hirata, J. Chem. Phys. 121, 51 (2004).
${ }^{39}$ M. Kállay and J. Gauss, J. Chem. Phys. 121, 9257 (2004).
${ }^{40}$ C. F. Bender and E. R. Davidson, Phys. Rev. 183, 23 (1969).
${ }^{41}$ J. L. Whitten and M. Hackmeyer, J. Chem. Phys. 51, 5584 (1969).
${ }^{42}$ B. Huron, J. P. Malrieu, and P. Rancurel, J. Chem. Phys. 58, 5745 (1973).
${ }^{43}$ G. H. Booth, A. J. W. Thom, and A. Alavi, J. Chem. Phys. 131, 054106 (2009).
${ }^{44}$ J. E. Deustua, J. Shen, and P. Piecuch, Phys. Rev. Lett. 119, 223003 (2017).
${ }^{45}$ A. A. Holmes, N. M. Tubman, and C. J. Umrigar, J. Chem. Theory Comput. 12, 3674 (2016).
${ }^{46}$ J. J. Eriksen, F. Lipparini, and J. Gauss, J. Phys. Chem. Lett. 8, 4633 (2017).
${ }^{47}$ E. Xu, M. Uejima, and S. L. Ten-no, Phys. Rev. Lett. 121, 113001 (2018).
${ }^{48}$ E. Giner, A. Scemama, and M. Caffarel, Can. J. Chem. 91, 879 (2013).
${ }^{49}$ E. Giner, A. Scemama, and M. Caffarel, J. Chem. Phys. 142, 044115 (2015).
${ }^{50}$ F. A. Evangelista, J. Chem. Phys. 140, 124114 (2014).
${ }^{51}$ N. M. Tubman, J. Lee, T. Y. Takeshita, M. Head-Gordon, and K. B. Whaley, J. Chem. Phys. 145, 044112 (2016).
${ }^{52}$ W. Liu and M. R. Hoffmann, J. Chem. Theory Comput. 12, 1169 (2016).
${ }^{53}$ Y. Ohtsuka and J.-y. Hasegawa, J. Chem. Phys. 147, 034102 (2017).
${ }^{54}$ P. M. Zimmerman, J. Chem. Phys. 146, 104102 (2017).
${ }^{55}$ J. P. Coe, J. Chem. Theory Comput. 14, 5739 (2018).
${ }^{56}$ Y. Garniron, A. Scemama, E. Giner, M. Caffarel, and P.-F. Loos, J. Chem. Phys. 149, 064103 (2018).
${ }^{57}$ J. J. Eriksen and J. Gauss, J. Phys. Chem. Lett. 10, 7910 (2019).
${ }^{58}$ A. A. Holmes, C. J. Umrigar, and S. Sharma, J. Chem. Phys. 147, 164111 (2017).
${ }^{59}$ J. Li, M. Otten, A. A. Holmes, S. Sharma, and C. J. Umrigar, J. Chem. Phys. 149, 214110 (2018).
${ }^{60}$ J. Li, Y. Yao, A. A. Holmes, M. Otten, Q. Sun, S. Sharma, and C. J. Umrigar, Phys. Rev. Res. 2, 012015 (2020).
${ }^{61}$ P.-F. Loos, A. Scemama, A. Blondel, Y. Garniron, M. Caffarel, and D. Jacquemin, J. Chem. Theory Comput. 14, 4360 (2018).
${ }^{62}$ A. D. Chien, A. A. Holmes, M. Otten, C. J. Umrigar, S. Sharma, and P. M. Zimmerman, J. Phys. Chem. A 122, 2714 (2018).
${ }^{63}$ P.-F. Loos, M. Boggio-Pasqua, A. Scemama, M. Caffarel, and D. Jacquemin, J. Chem. Theory Comput. 15, 1939 (2019).
${ }^{64}$ P.-F. Loos, F. Lipparini, M. Boggio-Pasqua, A. Scemama, and D. Jacquemin, J. Chem. Theory Comput. 16, 1711 (2020).
${ }^{65}$ P.-F. Loos, A. Scemama, M. Boggio-Pasqua, and D. Jacquemin, J. Chem. Theory Comput. 16, 3720 (2020).
${ }^{66}$ P.-F. Loos, Y. Damour, and A. Scemama, J. Chem. Phys. 153, 176101 (2020).
${ }^{67}$ Y. Garniron, K. Gasperich, T. Applencourt, A. Benali, A. Ferté, J. Paquier, B. Pradines, R. Assaraf, P. Reinhardt, J. Toulouse, P. Barbaresco, N. Renon, G. David, J.-P. Malrieu, M. Véril, M. Caffarel, P.-F. Loos, E. Giner, and A. Scemama, J. Chem. Theory Comput. 15, 3591 (2019).
${ }^{68}$ J. J. Eriksen, T. A. Anderson, J. E. Deustua, K. Ghanem, D. Hait, M. R. Hoffmann, S. Lee, D. S. Levine, I. Magoulas, J. Shen, N. M. Tubman, K. B. Whaley, E. Xu, Y. Yao, N. Zhang, A. Alavi, G. K.-L. Chan, M. Head-Gordon, W. Liu, P. Piecuch, S. Sharma, S. L. Ten-no, C. J. Umrigar, and J. Gauss, J. Phys. Chem. Lett. 11, 8922 (2020).
${ }^{69}$ Y. Yao, E. Giner, J. Li, J. Toulouse, and C. J. Umrigar, "Almost exact energies for the Gaussian-2 set with the semistochastic heat-bath configuration interaction method," J. Chem. Phys. 153, 124117 (2020).
${ }^{70}$ M. Véril, A. Scemama, M. Caffarel, F. Lipparini, M. Boggio-Pasqua, D. Jacquemin, and P.-F. Loos, Wiley Interdiscip. Rev.: Comput. Mol. Sci. 11, e1517 (2021).
${ }^{71}$ P.-F. Loos, A. Scemama, and D. Jacquemin, J. Phys. Chem. Lett. 11, 2374 (2020). ${ }^{72}$ J. J. Eriksen, J. Phys. Chem. Lett. 12, 418 (2021).
${ }^{73}$ Y. Garniron, A. Scemama, P.-F. Loos, and M. Caffarel, J. Chem. Phys. 147, 034101 (2017).
${ }^{74}$ S. Sharma, A. A. Holmes, G. Jeanmairet, A. Alavi, and C. J. Umrigar, J. Chem. Theory Comput. 13, 1595 (2017).
${ }^{75}$ D. A. Matthews, J. Chem. Theory Comput. 16, 6195 (2020).
${ }^{76}$ D. A. Matthews, L. Cheng, M. E. Harding, F. Lipparini, S. Stopkowicz, T.-C. Jagau, P. G. Szalay, J. Gauss, and J. F. Stanton, J. Chem. Phys. 152, 214108 (2020).
${ }^{77}$ M. Kállay, P. R. Nagy, D. Mester, Z. Rolik, G. Samu, J. Csontos, J. Csóka, P. B. Szabó, L. Gyevi-Nagy, B. Hégely, I. Ladjánszki, L. Szegedy, B. Ladóczki, K. Petrov, M. Farkas, P. D. Mezei, and A. Ganyecz, J. Chem. Phys. 152, 074107 (2020).
${ }^{78}$ D. A. Matthews and J. F. Stanton, J. Chem. Phys. 142, 064108 (2015).
${ }^{79}$ E. R. Davidson, J. Comput. Phys. 17, 87 (1975).
${ }^{80}$ A. Chrayteh, A. Blondel, P.-F. Loos, and D. Jacquemin, J. Chem. Theory Comput. 17, 416 (2021).
${ }^{81}$ N. T. Maitra, F. Zhang, R. J. Cave, and K. Burke, J. Chem. Phys. 120, 5932 (2004).
${ }^{82}$ B. Saha, M. Ehara, and H. Nakatsuji, J. Chem. Phys. 125, 014316 (2006).
${ }^{83}$ M. A. Watson and G. K.-L. Chan, J. Chem. Theory Comput. 8, 4013 (2012).
${ }^{84}$ Y. Shu and D. G. Truhlar, J. Am. Chem. Soc. 139, 13770 (2017).
${ }^{85}$ G. M. J. Barca, A. T. B. Gilbert, and P. M. W. Gill, J. Chem. Theory Comput. 14, 1501 (2018).
${ }^{86}$ P.-F. Loos, A. Devin, and D. Filippo, (2021). "How accurate are EOMCC4 vertical excitation energies?," Zenodo Version 1.0. http://doi.org/10.5281/ zenodo. 4739288

